

Specimen banking of passive samplers for aquatic environments: purpose, possibilities and procedures

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Specimen banking of passive samplers (PS)?

No specimen banks currently include PS!

Surely passive samplers has to be the optimum matrix for specimen banking!

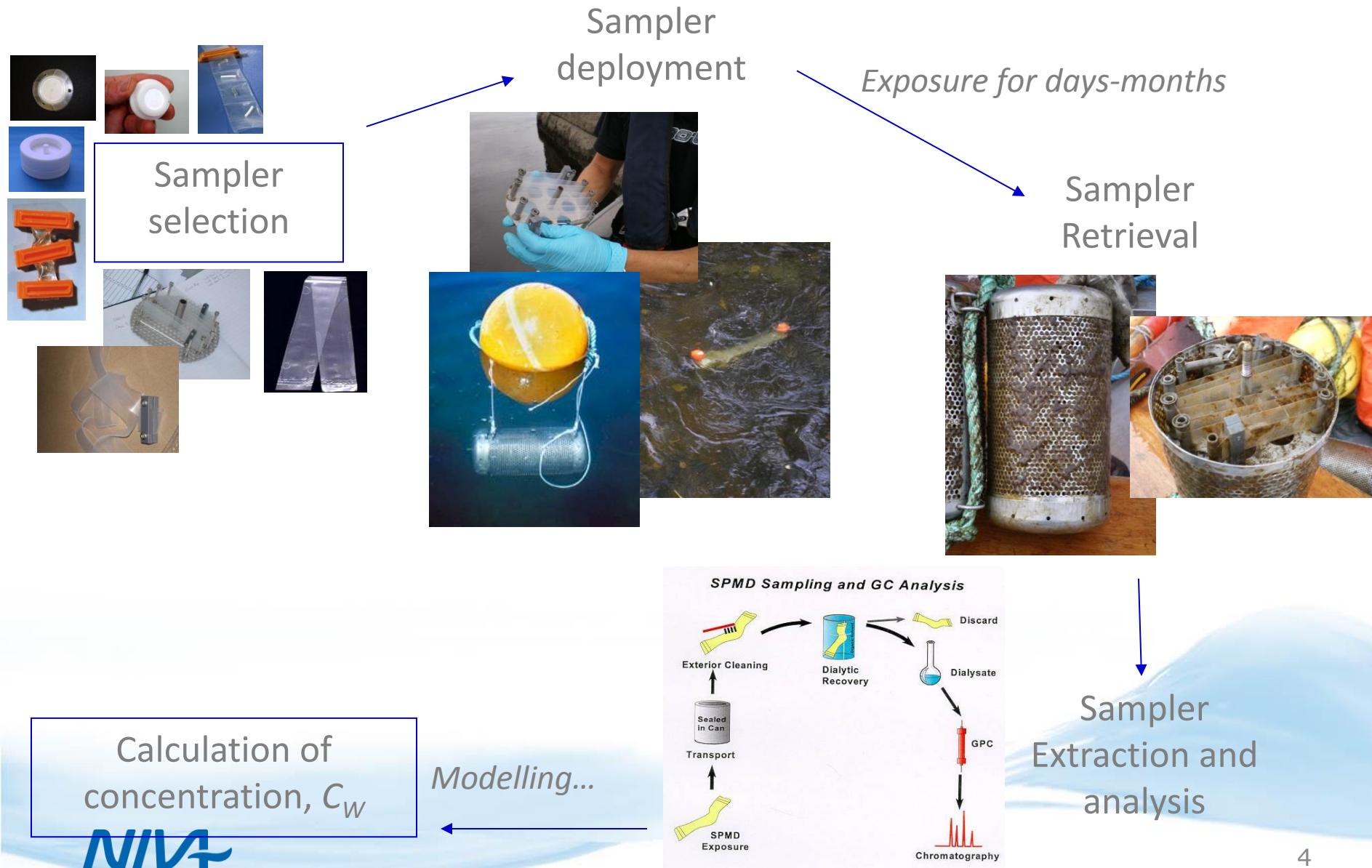
why?

- Difficult to specimen bank water, easier with biota or sediment but still requiring manipulation without blanks
- Possibility to use proper blanks/control samplers, so better traceability
- Don't take much space in the freezer

What's different from «regular» SB matrices?

- PS is not a environmental matrix but a manufactured device
 - Selectivity from the onset of PS
 - How much chemical is in the sampler after exposure depends on the sampler configuration and sampling operation
- The availability of blanks and preparation and field control samplers
 - Almost identical treatment of control and exposed samplers can be achieved (preparation, deployment/retrieval, surface cleaning and storage)
 - More than one control sampler for every exposed sampler may be needed
- The possibility to use «reference materials» and spiked polymers for stability studies
- Requires additional information to estimate an environmental concentration
 - Most robust procedure is to obtain this information for each individual sampler
 - Requires the user to estimate a sampling rate unless only qualitative data is needed
- Low volume and ease of storage

Principle of passive sampling (in water)



Sampler deployment & retrieval

- Standard equipment
- Contamination
- Deployment procedure

Sampler storage & transport

- Stability
- Temperature
- PRC/analyte stability
- Contamination (cross-)

Environmental variability

- (Bio)fouling
- Temperature
- Salinity
- Sunlight

Sampler/polymer calibration

- Calibration rig set-up (tank, carousel, analyte delivery)
- Polymer-water partition coefficients estimation (K_{pw})
- Polymer diffusion coefficient estimation (D)

*TRUE
concentration*

*Passive sampler-estimated
concentration*

Sampler selection

- Membrane thickness
- Surface/volume
- PRC concentration
- Contamination

Sampler extraction

- Contamination
- Recoveries

Analysis for analytes and PRCs

- Calibration, linearity
- LOD/LOQ ... etc

Procedure to estimate R_s and C_{free}

- R_s estimation procedure
- Model selection
- K_{pw} values etc etc...

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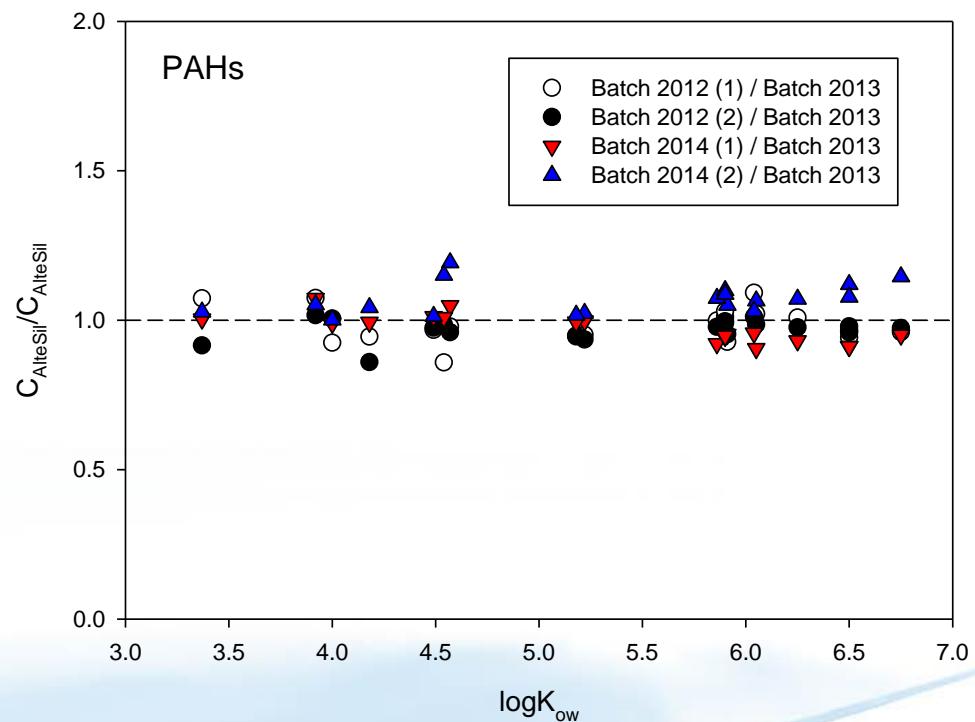
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Polymer selection for hydrophobic contaminants

Polymer selection (absorption-based passive sampling):

- Availability, practicality and cost, consistency of the supply

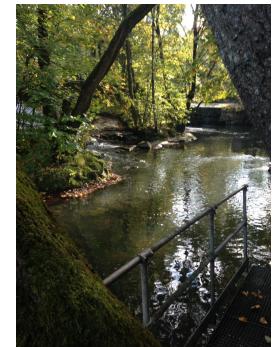
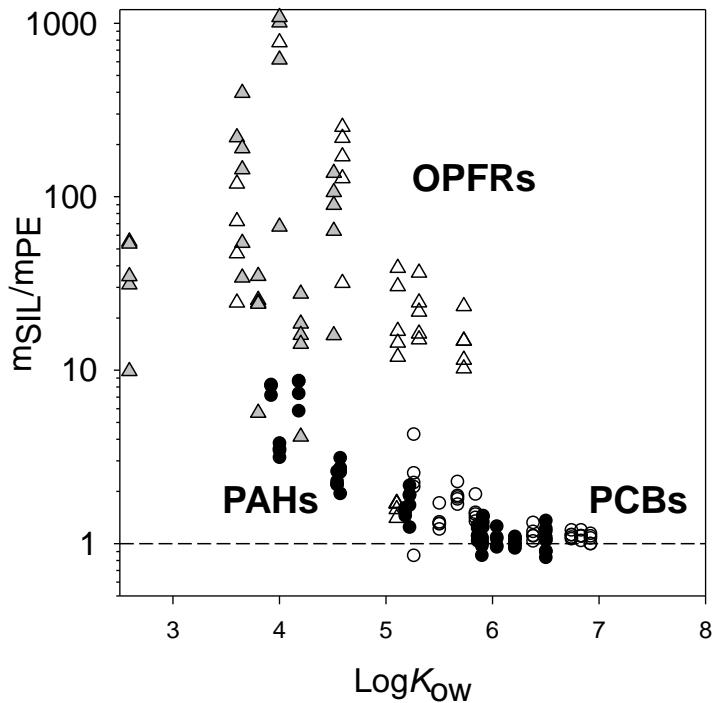
AlteSil™ silicone rubber sheets from different batches equilibrated in 60:40 methanol-water solution



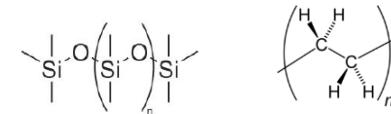
Polymer selection for hydrophobic contaminants

Polymer selection (absorption-based passive sampling):

- Sufficiently permeable for compounds of interest
- High pre-concentration factor (high sampler-water partition coefficient)



River Alna



	Silicone rubber	LDPE
Mass (g)	8.5	1.1
Surface area (cm ²)	300	300

*Allan et al., 2013. Passive sampling for target and nontarget analyses of moderately polar and nonpolar substances in water. *Environmental Toxicology and Chemistry* 32, 1718-1726.

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Calibration for emerging contaminants

- 33 chemicals of emerging concern (CECs)
- Focus on using single phase samplers (silicone rubber and LDPE)
- Polymer calibration*:
 - Polymer diffusion coefficients, D_p with a film stack experiment
 - Polymer-water partition coefficients, K_{pw} measurement with validation using a co-solvent method

*Pintado-Herrera, Lara-Martín, González-Mazo, and Allan. Determination of silicone rubber and low density polyethylene diffusion and polymer-water partition coefficients for emerging contaminants. *In press in ET&C, 2015*

Calibration data for 33 CECs

Example of measurement in the River Alna (Oslo)

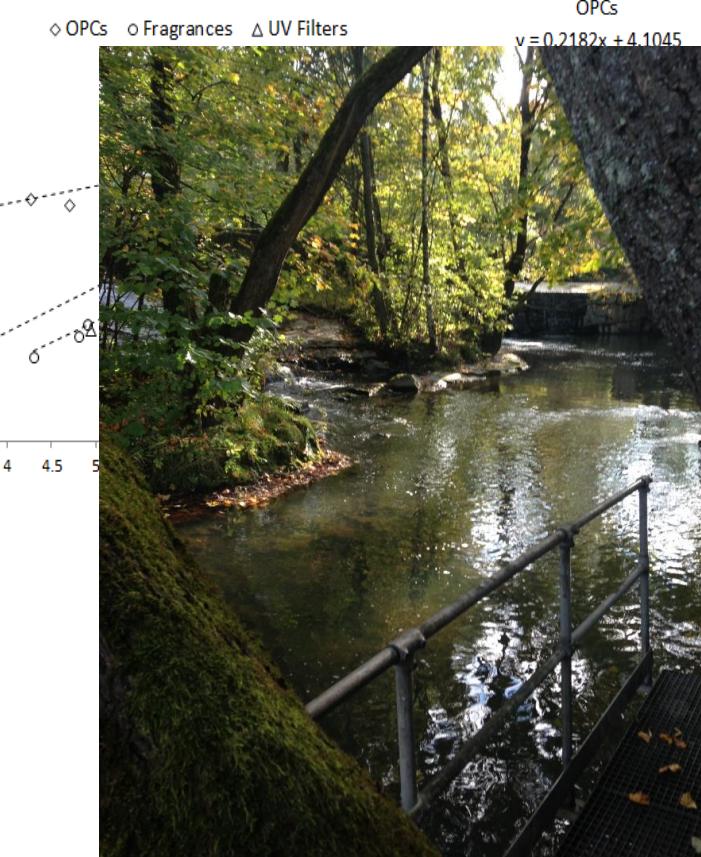
$\log D_p$ $\log K_{pw}$

A

Chemical	Silicone rubber	MW	C_{free} (ng L ⁻¹)				
	0	100	200	300	400	500	
Galaxolide	Synthetic musk		1600				
Tonalide	Synthetic musk		51				
Oxybenzone	UV filter		108				
Octocrylene	UV filter		448				
Triclosan	Antibacterial		9				
2-ethylhexyl-4-methoxycinnamate	UV filter		4				
Triphenylphosphate	Flame retardant/plasticiser		170				

B

$\log D_p$



- ▲ Endocrine disruptor compounds
- UV filters
- ◆ Fragrances
- OPCs
- UV filters out-of-range
- OC pesticides (Hale et al. 2010)
- + OPCs lower detection limits

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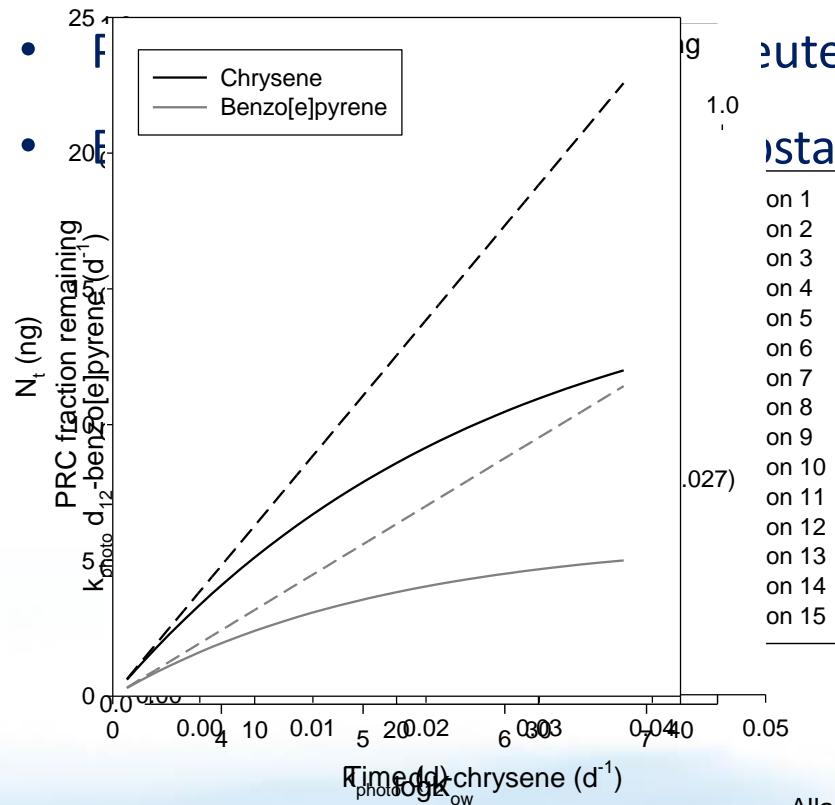
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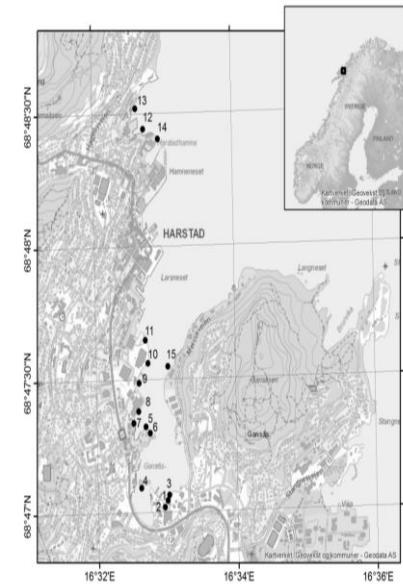
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Effect of environmental conditions: Photodegradation

- Semipermeable membrane devices (SPMDs)
- Exposure for 36 days in May at 15 locations in Harstad
- Performance Reference Compounds: Deuterated PAHs



Deuterated PAH PRCs
15 instances being sampled



Allan, I. J., Christensen, G., Bæk, K., & Evensen, A. (2016). Photodegradation of PAHs in passive water samplers. *Marine Pollution Bulletin*, in press

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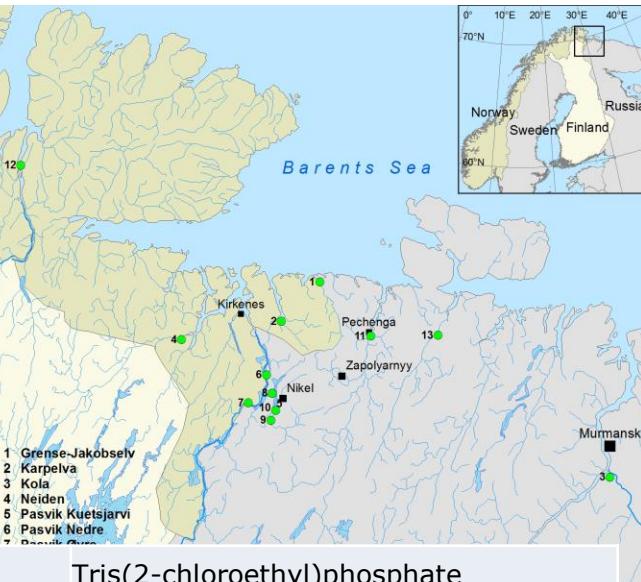
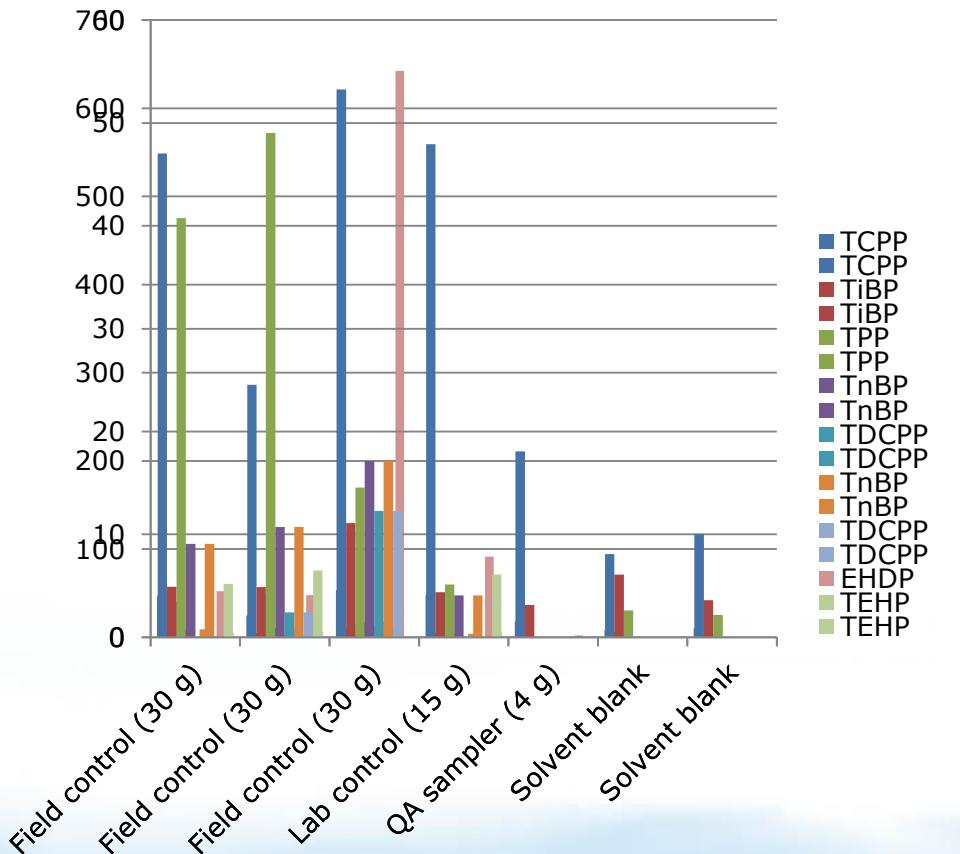
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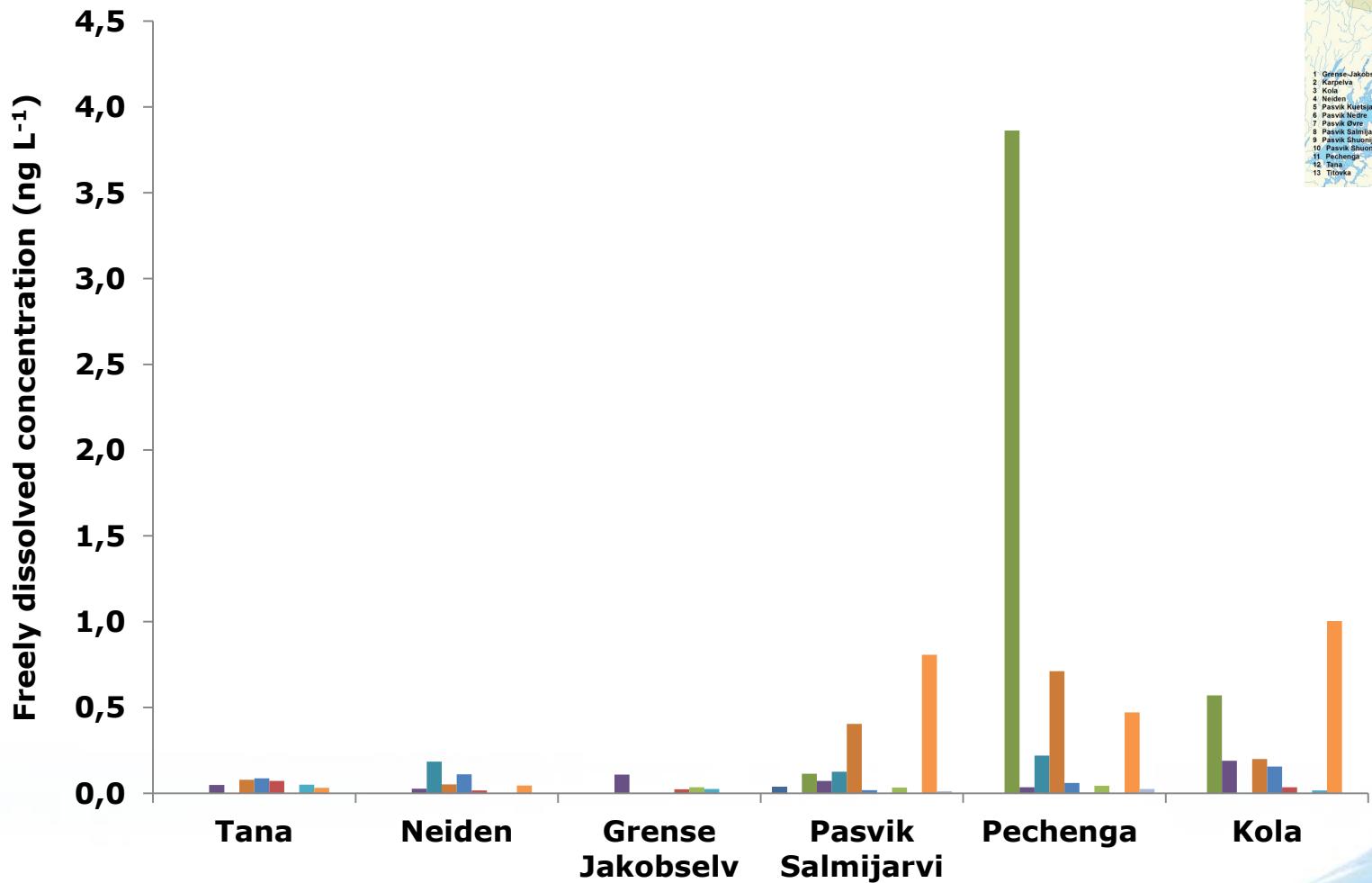
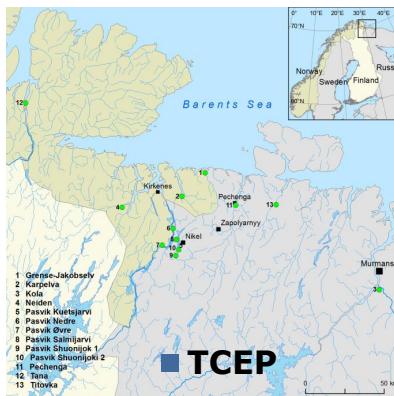
Contamination: Need for blanks

- Silicone rubber samplers (30 g, 1000 cm²) deployed in Norwegian and Russian Arctic
- OPFRs in QA samplers (ng sampler⁻¹)



TCEP	Tris(2-chloroethyl)phosphate
TPrP	Tripropylphosphate
TCPP	Tris(chloro iso-propyl)phosphate
TiBP	Tri-iso-butylphosphate
BdPhP	Butyldiphenylphosphate
TPP	Triphenylphosphate
DBPhP	Dibutylphenylphosphate
TnBP	tributylphosphate
TDCPP	Tris(1,3-dichloro-2-propyl)phosphate
TBEP	Tri(2-butoxyethyl)phosphate
TCP	Tricresylphosphate
EHDP	2-ethylhexyl-di-phenylphosphate
TEHP	Tris(2-ethylhexyl)phosphate
DCP	diphenylcresylphosphate
T35DMPP	Tris(3,5-dimethylphenyl) phosphate
TDBrPP	Tris(2,3-dibrompropyl) phosphate

OPFRs in Russian and Norwegian Arctic rivers



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Passive sampling in water, sediment, biota or humans

Where are we now?

	Water	Sediment	Biota	Human
Robustness	***	***	*	*
Validation needs	*	*	***	***
Metadata	***	**	***	*
Info on sampling kinetics	***	**	**	*
Need for blanks	***	***	***	**
Contamination				
-Field manipulation	***	**	**	*
-Lab manipulation	*	***	***	*

* Low

*** High

Passive sampling Cross-Working Group Activity activity for 2016

Objective:

- To develop a common repository for passive sampling data
- To prepare specifications for a module to input passive sampling data into NORMAN's EMPODAT database
- Two teleconference meetings before this summer
- A working meeting this autumn

Concluding remarks

- Specimen banking of PS is feasible
- Can provide a solution to specimen banking of complex matrices
- Robustness of the method: use of blanks and control samplers
- PS has «built-in» selectivity of the chemicals that are sampled and that can be looked for retrospectively
- Requires perhaps more data collection/metadata than the more common specimen bank matrices?
- Needs additional analyses and calculations to estimate environmental concentrations

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